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10/618,111 07/11/2003		James D.B. Smith	2003P08574US 4106			
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Siemens Corporation			FEELY, MICHAEL J			
Intellectual Pro	perty Department					
170 Wood Avenue South			ART UNIT	PAPER NUMBER		
Iselin, NJ 08830			1712			

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Please find below and/or attached an Office communication concerning this application or proceeding.

		Application N	D.	Applicant(s)				
		10/618,111		SMITH, JAMES D.B.				
	Office Action Summary	Examiner		Art Unit				
		Michael J. Fee		1712				
Period fo	The MAILING DATE of this communication app or Reply	pears on the cov	er sheet with the c	orrespondence ad	Idress			
THE - External after - If the - If NC - Failu Any	ORTENED STATUTORY PERIOD FOR REPLY MAILING DATE OF THIS COMMUNICATION. SIX (6) MONTHS from the mailing date of this communication. Period for reply specified above is less than thirty (30) days, a reply period for reply is specified above, the maximum statutory period or re to reply within the set or extended period for reply will, by statute reply received by the Office later than three months after the mailing ed patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, ho y within the statutory r will apply and will expi e, cause the application	wever, may a reply be tim ninimum of thirty (30) days re SIX (6) MONTHS from to to become ABANDONED	ely filed s will be considered timel the mailing date of this c O (35 U.S.C. § 133).				
Status								
1)⊠ 2a)□ 3)□	Responsive to communication(s) filed on <u>11 July 2003</u> . This action is FINAL . 2b) This action is non-final. Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.							
Disposit	ion of Claims							
5)□ 6)⊠ 7)□	Claim(s) 1-19 is/are pending in the application. 4a) Of the above claim(s) is/are withdrawn from consideration. Claim(s) is/are allowed. Claim(s) 1-19 is/are rejected. Claim(s) is/are objected to. Claim(s) is/are objected to. Claim(s) are subject to restriction and/or election requirement.							
Applicat	ion Papers							
10)⊠	The specification is objected to by the Examine The drawing(s) filed on 11 July 2003 is/are: a) Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct The oath or declaration is objected to by the Examine The specific and the spec	i⊠ accepted or drawing(s) be he tion is required if	ld in abeyance. See the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 C				
Priority	under 35 U.S.C. § 119							
12)□ a)	Acknowledgment is made of a claim for foreign All b) Some * c) None of: 1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority application from the International Burea See the attached detailed Office action for a list	ts have been re ts have been re prity documents tu (PCT Rule 17	ceived. ceived in Applicati have been receive (.2(a)).	on No ed in this National	l Stage			
2) Notice 3) Information	ce of References Cited (PTO-892) ce of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO-1449 or PTO/SB/08) er No(s)/Mail Date 0703.) 5)	Interview Summary Paper No(s)/Mail Da Notice of Informal F Other:	ate	⁻ O-152)			

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DETAILED ACTION

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. Claims 1-9 and 11-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Smith et al. (US Pat. No. 5,904,984) in view of Cook et al. (US Pat. No. 6,369,183).

Regarding claims 1-9 and 11, Smith et al. disclose: (1) an LCT-epoxy polymer (column 2, line 22 through column 5, line 19); (7) further comprising a process step of mixing at least one anhydriding agent with said LCT-epoxy polymer (column 5, lines 20-56); (8) wherein said anhydriding agent is taken from the group consisting of 1-methylhexahydrophthalic anhydride and 1-methyltetrahydrophthalic anhydride (column 5, lines 35-41); (9) wherein said anhydriding agent is approximately 25-45% by weight of said LCT-epoxy polymer (column 5, lines 51-53); and (11) wherein said mixture is added to an electrical insulator as coating before curing (Abstract; column 2, lines 22-29; Figure 2).

The composition of Smith et al. also contains inorganic particulate filler, including aluminum-based fillers, to improve the electrical properties of the resin (column 6, lines 45-50). However, the teachings of Smith et al. are deficient in that they do not teach: (1) a method of making a homogeneous *alumoxane-LCT-epoxy polymer* comprising:

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Mixing at least one boehmite material with the LCT-epoxy resin under conditions to form
a uniform dispersion and an essentially complete co-reactivity of the boehmite material
with the LCT-epoxy resin;

- Curing the mixture to produce the homogeneous *alumoxane-LCT-epoxy polymer*;
 - Wherein the weight ratio of boehmite material to LCT-epoxy resin is between
 3:17 and 13:7 (15-65% boehmite base on combined weight of boehmite and
 LCT-epoxy resin); and
 - Wherein the alumoxane-LCT-epoxy polymer has a dielectric strength of at least 1.2 kV/mil;
- (2) wherein the boehmite material comprises carboxylate-alumoxane; (3) wherein the carboxylate-alumoxane is 4-hydroxybenoate-alumoxane; (4) wherein the alumoxane portions of the alumoxane-LCT-epoxy polymer is 20-50% by weight; (5) wherein the mixing step comprises first pre-heating the LCT-epoxy resin until it is clear; and (6) wherein the mixing step further comprises warming until said mixture is clear.

Cook et al. disclose a method of forming hybrid polymers and resins in which at least one component is a chemically modified carboxylate-alumoxane. The carboxylate-alumoxanes are chemically bonded into the polymer backbone through reaction of the appropriate functional groups of a polymer precursor with the carboxylate-alumoxane (see column 1, lines 14-21). This method is an alternative to the standard practice of adding filler to resin systems in order to enhance properties, such as thermal conductivity. This method also avoids handling and dispersion problems, including phase separation and agglomeration of the filler in the polymer composite (see column 1, line 52 through column 2, line 54). This method can be applied to

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numerous polymer materials, including epoxy resins. The reaction mechanism can be found in Figure 10. Essentially, Cook et al. disclose: (1) a method of making a homogeneous alumoxane-epoxy polymer comprising:

- Mixing at least one boehmite material with the LCT-epoxy resin under conditions to form
 a uniform dispersion and an essentially complete co-reactivity of the boehmite material
 with the LCT-epoxy resin (column 11, line 52 through column 12, line 51; Figures 10 &
 11);
- Curing the mixture to produce the homogeneous *alumoxane-epoxy polymer* (column 11, lines 29-51; Figures 7-9);
 - O Wherein the weight ratio of boehmite material to LCT-epoxy resin is between 3:17 and 13:7 (column 14, lines 17-19; Table 1: column 25, lines 55-64);

(2) wherein the boehmite material comprises carboxylate-alumoxane (column 11, line 29 through column 12, line 10); (3) wherein the carboxylate-alumoxane is 4-hydroxybenoate-alumoxane (column 11, line 65); (4) wherein the alumoxane portions of the alumoxane-epoxy polymer is 20-50% by weight (column 14, lines 17-19; Table 1: column 25, lines 55-64); (5) wherein the mixing step comprises first pre-heating the LCT-epoxy resin until it is clear (column 12, lines 24-51); and (6) wherein the mixing step further comprises warming until said mixture is clear (column 12, lines 24-51).

Cook et al. do not mention the use of LCT-epoxy polymers; however, they disclose, "Although we have demonstrated the formation of alumoxane-epoxy composites using commercial resins such as Dow Chemical's DER 332, combinations of DER 332 and DER 732, Union Carbide's ERL (cycloaliphatic resin), and Shell's EPI-REZ (a waterborne resin system),

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any commercially available epoxy resin can be used to prepare the carboxylate-

alumoxane/epoxy polymer hybrid materials," (column 12, lines 11-17). In light of this, LCT-epoxy polymers would fall under this umbrella of suitable materials – see MPEP 2144.07. Furthermore, one skilled in the art would have been motivated to employ this method in the method of Smith et al. in order to avoid the drawbacks of physically adding filler materials, especially aluminum-based filler materials, to their LCT-epoxy polymer compositions.

Lastly, the combined teachings of Smith et al. and Cook et al. are silent regarding dielectric strength of the alumoxane-LCT-epoxy polymers. However, it appears that if one skilled in the art were to apply the technique of Cook et al. using the LCT-epoxy polymers of Smith et al., this property would have been an inherent result of the reaction between the carboxylate-alumoxane and the LCT-epoxy polymer – *see MPEP 2112.01 II*.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to employ the technique of reacting a carboxylate-alumoxane with an epoxy resin to form a hybrid epoxy resin, as taught by Cook et al., with the LCT-epoxy resins of Smith et al. because the teachings of Cook et al. extend to all commercially available epoxy resins, resulting in a hybrid polymer having a desired set of properties provided by the alumoxane. This technique is an alternative to the standard practice of adding filler to resin systems and it also avoids handling and dispersion problems of filler materials, including phase separation and agglomeration of filler in the polymer composite.

Regarding claims 12-14, the combined teachings of Smith et al. and Cook et al. are as set forth above and incorporated herein. The scope of claims 12-14 is similar to the scope of claims 1-9 and 11; however, it further contains language regarding the steps of coating and

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impregnating the resin onto/into an electrical insulator, wherein the electrical insulator is a mica/glass insulating tape. The base reference of Smith et al. teaches these limitations (see column 8, line 49 through column 10, line 7; and claim 3).

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to employ the technique of reacting a carboxylate-alumoxane with an epoxy resin to form a hybrid epoxy resin, as taught by Cook et al., with the LCT-epoxy resins of Smith et al. because the teachings of Cook et al. extend to all commercially available epoxy resins, resulting in a hybrid polymer having a desired set of properties provided by the alumoxane. This technique is an alternative to the standard practice of adding filler to resin systems and it also avoids handling and dispersion problems of filler materials, including phase separation and agglomeration of filler in the polymer composite.

Regarding claims 15-19, the combined teachings of Smith et al. and Cook et al. are as set forth above and incorporated herein. The scope of claims 15-19 is similar to the scope of claims 1-9 and 11; however, it further contains the following properties:

- thermal conductivity in the transverse direction of at least 0.50 W/mK and in the thickness direction of at least 0.99 W/mK in an environment of 25°C; and
- the alumoxane-LCT-epoxy polymers are substantially free of particle wetting and micro-void formation.

The combined teachings of Smith et al. and Cook et al. are silent regarding these properties of the alumoxane-LCT-epoxy polymers. However, it appears that if one skilled in the art were to apply the technique of Cook et al. using the LCT-epoxy polymers of Smith et al., these

properties would have been an inherent result of the reaction between the carboxylate-alumoxane and the LCT-epoxy polymer – see MPEP 2112.01 II.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to employ the technique of reacting a carboxylate-alumoxane with an epoxy resin to form a hybrid epoxy resin, as taught by Cook et al., with the LCT-epoxy resins of Smith et al. because the teachings of Cook et al. extend to all commercially available epoxy resins, resulting in a hybrid polymer having a desired set of properties provided by the alumoxane. This technique is an alternative to the standard practice of adding filler to resin systems and it also avoids handling and dispersion problems of filler materials, including phase separation and agglomeration of filler in the polymer composite.

3. Claim 10 is rejected under 35 U.S.C. 103(a) as being unpatentable over Smith et al. (US Pat. No. 5,904,984) and Cook et al. (US Pat. No. 6,369,183) in view of Stackhouse et al. (US Pat. No. 4,427,740).

Regarding claim 10, the combined teachings of Smith et al. and Cook et al. are as set forth above and incorporated herein. The combined teachings are silent regarding the addition of zinc naphthenate or chromium acetylacetonate.

Stackhouse et al. disclose an insulated electrical member adapted for high voltage-use, wherein the material is impregnated with an epoxy resin (Abstract; column 6, lines 50-56). They further disclose, "Typical reactive impregnating compositions comprise, for example, bisphenol A epoxy resin; acid anhydride, curing agent, such as NADIC methyl anhydride or 1-methyltetrahydrophthalic anhydride; reactive diluent, such as neopentyl diglycidyl ether or vinyl

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cyclohexanone dioxide; and latent accelerator, such as chromium acetylacetonate, to produce a low viscosity composition of about 80 cps at 25°C," (column 6, line 67 through column 7, line 6).

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The teachings of Stackhouse et al. demonstrate that chromium acetylacetonate is recognized in the art as a suitable latent accelerator for epoxy/anhydride systems used to impregnate electrical insulating materials. In light of this, it has been found that the selection of a known material based on its suitability for its intended use supports a *prima facie* obviousness determination – *see MPEP 2144.07*.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to add a latent catalyst of zinc naphthenate or chromium acetylacetonate to the combined teachings of Smith et al. and Cook et al. because Stackhouse et al. demonstrate that chromium acetylacetonate is recognized in the art as a suitable latent accelerator for epoxy/anhydride systems used to impregnate electrical insulating materials.

Communication

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Michael J. Feely whose telephone number is 571-272-1086. The examiner can normally be reached on M-F 8:30 to 5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Randy Gulakowski can be reached on 571-272-1302. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Michael J. Feely Patent Examiner Art Unit 1712

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